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Harmon Craig

Harmon Craig (March 15, 1926 – March 14, 2003) was an American geochemist who worked briefly for the University of Chicago (1951-1955) before spending the majority of his career at Scripps Institution of Oceanography (1955-2003).^[1]

Craig was involved in numerous research expeditions, which visited the Great Rift Valley of East Africa,^[2] the crater of Loihi (now known as Kamaʻehuakanaloa), the Afar Depression of Ethiopia, Greenland's ice cores, and Yellowstone's geysers, among many others.^[3] This led to him being described as "the Indiana Jones of the Earth sciences",^[4] someone "whose overriding impulse was to get out and see the world they were studying".^[3]

Craig made many significant discoveries in geochemistry. He is credited with establishing the field of carbon isotope geochemistry by characterizing carbon's stable isotopic signatures in various natural materials.^[5] This had immediate applications in radiocarbon dating.^{[1]:4–5} By studying stable and radioactive carbon isotopes in the biosphere and air-sea system, he derived the atmospheric residence time of carbon dioxide with respect to oceanic uptake. His work laid the foundation for isotopic studies of the carbon cycle, and was fundamental to understanding carbon sequestering in the oceanic and the terrestrial biosphere and the modulation of global warming.^{[5][6][7]} In addition, from 1969 to 1989, Harmon Craig served as an editor for *Earth and Planetary Science Letters*.^[8]

Family and early life

Harmon Craig was born March 15, 1926 in Manhattan, in New York City,^[1] to John Richard Craig, Jr. (1896-1945) and his wife Virginia (Stanley) Craig.^{[9][10]} He was named after his uncle, Harmon Bushnell Craig (1895-1917), but does not use his middle name.^{[1]:5}

Harmon Craig's grandparents on his father's side were actors, directors and producers. During World War I, John Craig (1868-1931) and his wife, actress Mary Young, led the first professional American stock theater company to travel to France and entertain troops at the front. While they entertained the troops, their sons Harmon Bushnell Craig (1895-1917) and John Richard Craig, Jr. (1896-1945) served in the war effort.^[11] John Craig, Jr. received a French Croix de Guerre^{[1]:10} for his efforts as a second lieutenant of artillery, working with French 75s.^[11] Harmon Bushnell Craig died serving with an ambulatory corps run by the American Field Service, and was posthumously awarded the French Croix de Guerre.^[12]

Harmon Craig	
Born	March 15, 1926 <div>New York City, <u>New York</u></div>
Died	March 14, 2003 (aged 76) <div>La Jolla, <u>California</u></div>
Nationality	<u>American</u>
Known for	<u>Geochemistry</u>
Spouse	Valerie Craig (m. 1947)
Awards	<u>V. M. Goldschmidt Award</u> (1979) <div><u>Vetlesen Prize</u> (1987)<div><u>Balzan Prize</u> (1988)</div></div>
Scientific career	
Institutions	<u>Scripps Institution of Oceanography</u>
Doctoral advisor	<u>Harold C. Urey</u>

In November 1924,^[13] John Craig, Jr. married Virginia Stanley of Wichita, Kansas. They had three children: Harmon (named after his uncle), John Richard III (named after his father and grandfather), and Stanley Craig.^[9]

Harmon Craig's mother, Virginia Stanley, was descended from Quakers who helped found schools for freed slaves. His mother's involvement with the Quakers was a strong influence on Harmon Craig.^{[1]:5}

University of Chicago

Harmon Craig studied geology and chemistry at the University of Chicago. In 1944, he joined the U.S. Navy, serving as a communications and radar officer during World War II. After the war, he continued his education at University of Chicago, working with Nobel Laureate Harold Urey.^[4] Craig credits Urey with giving him valuable advice on how to choose scientific problems: "If you go into a project, it's got to be a scientific problem that has rooms that continue into other rooms."^[14]

Craig earned his Ph.D. in 1951,^{[4][2]} with *The geochemistry of the stable carbon isotopes*, a thesis on carbon isotope geochemistry.^{[15][16][17]} Craig created his thesis to find the measurement of ancient sea temperature. Craig used the carbon dioxide released from calcium carbonate fossils as a basis for future researches involving the carbon system. The masses of carbon dioxide that are produced by ¹⁸O and ¹⁶O were used to calculate respective masses. Craig's study of the carbon isotope produced corrections that deal with mass fractionation and radiocarbon ages. Craig's thesis work is considered a foundational accomplishment for its studies of ¹³ C and ¹² C in a wide range of natural materials, including everything from ocean water to the atmosphere; volcanic gases; plants, coal, diamonds, and petroleum; sediments, igneous rocks and meteorites.^[5] His theory has been applied to applications as varied as determining food chains and the identifying the sources of stone for ancient statues.^[1] Karl Turekian has stated that "Craig's 35-year-old dissertation is still the measure of all subsequent work in the field."^[3]

Craig joined the Enrico Fermi Institute at the University of Chicago as a research associate in 1951.^[2] In 1953, Urey and Craig published results showing that chondrites, meteors from the Solar System, did not have a single fixed composition, as had been assumed. After carrying out analyses of the chemical composition of hundreds of different meteorites, they reported that chondrites fell into two distinguishable groups, high iron (H) and low iron (L) chondrites. Their work "underscored the value of reliable chemical data" and led to significant improvements in data analysis in the field.^[18] It led to a better understanding of the materials and processes involved in forming planets.^{[19][20]}

Scripps Institution of Oceanography

In 1955 Harmon Craig was recruited to Scripps Institution of Oceanography by Roger Revelle.^{[1]:5} His laboratory at Scripps eventually contained five mass spectrometers, one of them a portable unit.^[21] As a professor of geochemistry and oceanography at Scripps, Craig developed new methods in radiocarbon dating and applied radioisotope and isotope distribution to various topics in marine-, geo-, and cosmochemistry. Craig produced fundamental findings about how the deep earth, oceans and atmosphere work.^[5]

During the 1950s Craig measured variations in the concentrations of hydrogen and oxygen isotopes in natural waters. In 1961, Craig identified the global meteoric water line, a linear relationship describing the occurrence of hydrogen and oxygen isotopes in terrestrial waters.^{[15]:344[2][22]} Craig also

established the oxygen isotope shift in geothermal and volcanic fluids, demonstrating that the water is meteoric. His discovery outlined the relation between rocks and water in geothermal systems.^{[23][24][22]}

In 1963, Craig received a Guggenheim Fellowship, using it to spend a year at the Istituto de Geologia Nucleare, Pisa, Italy. He described a framework for studying the isotopic composition of the hydrosphere, discussing kinetics, equilibrium, and the use of isotopes for paleoenvironmental reconstructions.^{[1]:6[24][25][26]} The work he presented with Louis I. Gordon on isotopic fractionation of the phase changes in water is known as the Craig-Gordon Model.^[27] The model is applied to problems in watershed and ecosystem studies such as the calculation of evaporation.^{[15]:355–358[25][28]} It has been called "a corner stone of isotope geochemistry."^[27]

During the Nova Expedition of 1967, Craig and colleagues W. Brian Clarke (1937–2002)^{[29]:449–450[30]} and M.A. Beg from McMaster University in Canada observed the Kermadec Trench in the Pacific Ocean. They found unexpectedly high proportions of the helium-3 isotope in the ocean waters. Craig concluded that the isotope was present within the Earth's mantle and theorized that it was leaking into sea water through cracks in the sea floor.^{[31][21][32]}

Craig and coworkers studied the isotopic composition of atmospheric and dissolved oxygen in the composition of dissolved gases, where he discovered the biochemical oxygen demand and the intake in the ocean mixed layer. Craig determined by measuring that the element, ²¹⁰Pb is rapidly scavenged by sinking particulate matter.^{[2][33][34]}

In 1970, Craig teamed up with colleagues at Scripps, Columbia University's Lamont–Doherty Earth Observatory and the Woods Hole Oceanographic Institution to direct the GEOSECS Programme (geochemical ocean sections study) to investigate the chemical and isotopic properties of the world's oceans.^[21] GEOSECS produced the most complete set of ocean chemistry data ever collected.^[2] In 1971, as part of the Antipode Expedition, Craig and his colleagues gathered hydrographic casts and other data, and discovered a benthic front separating the South Pacific deep and bottom water.^{[21]:338[1][35]}

During the 1970s Craig examined the relationship of gases such as radon and helium to earthquake prediction, developing a monitoring network at thermal springs and wells near major fault lines in southernmost California.^{[36][37]} In 1979, he detected an increase in radon and helium as a precursor to an earthquake near Big Bear Lake, California.^{[3][38][37]}

In a long-term project, Harmon Craig and Valerie Craig (his wife) used carbon and oxygen isotopes to identify the sources of the marble used in ancient Greek sculptures and temples.^{[39][40][41]}

Craig discovered submarine Hydrothermal vents by measuring helium 3 and radon emitted from seafloor spreading centers. He made 17 dives to the bottom of the ocean in the ALVIN submersible, including the first descent into the Mariana Trough. There he discovered hydrothermal vents nearly 3700m deep.^{[14][42]} Craig proved that there was excess ³He instead of ⁴He, affecting the understanding for ocean circulation and seafloor spreading.^{[43][44]}

Craig led 28 oceanographic expeditions and traveled to the East African Rift Valley, The Dead Sea, Tibet, Yunnan (China) and many other places to sample volcanic rocks and gases.^{[45][14]} He visited all the major volcanic island chains of the Pacific Ocean and Indian Ocean to collect lava samples. He identified 16 mantle hotspots where volcanic plumes rise from the Earth's outer core through the deep mantle by measuring their helium 3 to helium 4 ratio, identifying the higher helium 3 content present in the hotspots as primordial helium, trapped in the Earth's core when it was first formed.^[42]

Craig was one of the earliest people to analyze the gases trapped in the glacier ice.^{[46][1][47][48]} Craig reported that the methane in the atmosphere had increased twice due to human day-to-day activities in the last 300 years.^{[49][1][50][47]}

Awards and honors

Craig was elected to the National Academy of Sciences in 1979.^[51] Craig won the VM Goldschmidt Medal of the Geochemical Society in 1979, the National Science Foundation's Special Creativity Award in Oceanography in 1982 and the Arthur L. Day Prize and Lectureship of the National Academy of Sciences in 1987. He shared the Vetlesen Prize with Wallace S. Broecker in 1987.^{[49][52]}

In 1998 he was awarded the Balzan Prize for Geochemistry, from the International Balzan Foundation of Milan, Italy.^{[45][46][53]} The Foundation commended him as "a pioneer in earth sciences who uses the varied tools of isotope geochemistry to solve problems of fundamental scientific importance and immediate relevance in the atmosphere, hydrosphere and solid earth."^[2] It was the first time that the prize had gone to a geochemist. Craig was quoted as saying "The Prize's most significant effect was to establish that Geochemistry, especially Isotope Geochemistry, which began in 1947, had come of age and is a mature science. This was much more important than the specific person chosen for the award."^[42]

He received an honorary degree from the University of Paris.^{[1]:10}

Death

Craig died at Thornton Hospital in La Jolla, California on 14 March 2003^[2] from a massive heart attack^[5] a day before his seventy-seventh birthday.^[54]

Harmon's curiosity and sense of adventure knew no bounds... His drive for scientific achievement was unparalleled in my experience. The ocean and earth science world has lost a truly spirited adventurer and one of the greatest geochemists of the 20th century. – Charles Kennel, director of Scripps Institution of Oceanography, 2003^[2]

References

- Turekian, Karl K. (2006). "Harmon Craig, 1926-2003, a biographical memoir" (<http://www.nasonline.org/publications/biographical-memoirs/memoir-pdfs/craig-harmon.pdf>) (PDF). *Biography of the National Academy of Sciences*. National Academy of Sciences.
- Aguilera, Mario (March 18, 2003). "Obituary Notice Pioneer of Geochemistry: Harmon Craig" (<http://web.archive.org/web/20181122215548/https://scripps.ucsd.edu/news/2706>). *Scripps News*. Archived from the original (<https://scripps.ucsd.edu/news/2706>) on November 22, 2018. Retrieved November 22, 2018.
- Lawren, Bill (April 17, 1989). "Harmon Craig: Stalking Excellence, Leaving Controversy In His Wake" (<https://www.the-scientist.com/news/harmon-craig-stalking-excellence-leaving-controversy-in-his-wake-62103>). *The Scientist*. Retrieved 18 April 2019.
- Gates, Alexander E. (2009). *A to Z of Earth Scientists* (<https://books.google.com/books?id=iiG70wXOHMwC&pg=PA60>). Infobase Publishing. pp. 60–62. ISBN 9781438109190. Retrieved 18 April 2019.

5. Weiss, Ray (2003). "Harmon Craig (1926–2003)" (<https://doi.org/10.1029%2F2003EO220005>). *Eos, Transactions American Geophysical Union*. **84** (22): 207–208. Bibcode:2003EOSTr..84..207W (<https://ui.adsabs.harvard.edu/abs/2003EOSTr..84..207W>). doi:10.1029/2003EO220005 (<https://doi.org/10.1029%2F2003EO220005>).
6. Craig, Harmon (January 1957). "Isotopic standards for carbon and oxygen and correction factors for mass-spectrometric analysis of carbon dioxide". *Geochimica et Cosmochimica Acta*. **12** (1–2): 133–149. Bibcode:1957GeCoA..12..133C (<https://ui.adsabs.harvard.edu/abs/1957GeCoA..12..133C>). doi:10.1016/0016-7037(57)90024-8 (<https://doi.org/10.1016%2F0016-7037%2857%2990024-8>).
7. Craig, Harmon (February 1957). "The Natural Distribution of Radiocarbon and the Exchange Time of Carbon Dioxide Between Atmosphere and Sea". *Tellus*. **9** (1): 1–17. Bibcode:1957Tell....9....1C (<https://ui.adsabs.harvard.edu/abs/1957Tell....9....1C>). doi:10.1111/j.2153-3490.1957.tb01848.x (<https://doi.org/10.1111%2Fj.2153-3490.1957.tb01848.x>). S2CID 120862259 (<https://api.semanticscholar.org/CorpusID:120862259>).
8. "Accession No.: 2003-41 PROCESSING RECORD: Harmon Bushnell Craig Papers, 1948-2003" (<https://library.ucsd.edu/speccoll/findingaids/Craig2003-41.pdf>) (PDF). *SCRIPPS INSTITUTION OF OCEANOGRAPHY ARCHIVES*. Retrieved 19 April 2019.
9. "John Richard Craig, Jr" (<https://www.findagrave.com/memorial/182756196/john-richard-craig>). *Find A Grave*. Retrieved 18 April 2019.
10. "JOHN CRAIG JR., 47, A STAGE PRODUCER; Former Theatre Man Here and in Boston Dead on Coast-- Once in Insurance Field". *The New York Times*. December 6, 1945.
11. Evans, James W.; Harding, Gardner L. (1921). *Entertaining the American Army: The American Stage and Lyceum in the World War* (<https://archive.org/details/entertainingame01rundgoog>). New York: Association Press. pp. 66 (<https://archive.org/details/entertainingame01rundgoog/page/n92>)–73. ISBN 978-0353151208. Retrieved 18 April 2019.
12. Goldsmith, Louie. " "Serve my country to the last stitch": Honoring alumni lost in World War I" (<https://web.archive.org/web/20190418145958/https://thesagonline.com/22907/news/serve-my-country-to-the-last-stitch-honoring-alumni-lost-in-world-war-1/>). *The Sagamore*. Archived from the original (<https://thesagonline.com/22907/news/serve-my-country-to-the-last-stitch-honoring-alumni-lost-in-world-war-1/>) on 2019-04-18. Retrieved 2019-04-18.
13. "Wedding John Craig Jr and Virginia Stanley" (https://www.newspapers.com/clip/25714140/wedding_john_craig_jr_and_virginia/). *Daily News*. 7 Nov 1924. Retrieved 18 April 2019.
14. Sturchio, Neil (1999). "A conversation with Harmon Craig" (<https://www.geochemsoc.org/files/8313/4436/8118/gn098.pdf>) (PDF). *The Geochemical News*. No. January. pp. 12–20. Retrieved 18 April 2019.
15. Michener, Robert; Lajtha, Kate, eds. (2007). *Stable isotopes in ecology and environmental science* (2nd ed.). Malden, MA; Oxford, UK: Blackwell Publishing. pp. xx, 6–8. CiteSeerX 10.1.1.469.3198 (<https://citeseerx.ist.psu.edu/viewdoc/summary?doi=10.1.1.469.3198>). ISBN 978-1-4051-2680-9.
16. Craig, Harmon (1953). *The geochemistry of the stable carbon isotopes*. Chicago, Illinois: University of Chicago (Ph.D. Thesis).
17. Craig, Harmon (February 1953). "The geochemistry of the stable carbon isotopes". *Geochimica et Cosmochimica Acta*. **3** (2–3): 53–92. Bibcode:1953GeCoA...3...53C (<https://ui.adsabs.harvard.edu/abs/1953GeCoA...3...53C>). doi:10.1016/0016-7037(53)90001-5 (<https://doi.org/10.1016%2F0016-7037%2853%2990001-5>).
18. Jarosewich, Eugene (December 1990). "Chemical analyses of meteorites: A compilation of stony and iron meteorite analyses". *Meteoritics*. **25** (4): 323–337. Bibcode:1990Metic..25..323J (<https://ui.adsabs.harvard.edu/abs/1990Metic..25..323J>). doi:10.1111/j.1945-5100.1990.tb00717.x (<https://doi.org/10.1111%2Fj.1945-5100.1990.tb00717.x>).

19. Wiik, H.B. (June 1956). "The chemical composition of some stony meteorites". *Geochimica et Cosmochimica Acta*. **9** (5–6): 279–289. Bibcode:1956GeCoA...9..279W (<https://ui.adsabs.harvard.edu/abs/1956GeCoA...9..279W>). doi:10.1016/0016-7037(56)90028-X (<https://doi.org/10.1016%2F0016-7037%2856%2990028-X>).
20. Urey, Harold C.; Craig, Harmon (August 1953). "The composition of the stone meteorites and the origin of the meteorites". *Geochimica et Cosmochimica Acta*. **4** (1–2): 36–82. Bibcode:1953GeCoA...4...36U (<https://ui.adsabs.harvard.edu/abs/1953GeCoA...4...36U>). doi:10.1016/0016-7037(53)90064-7 (<https://doi.org/10.1016%2F0016-7037%2853%2990064-7>).
21. Shor, Elizabeth Noble (1978). *Scripps Institution of Oceanography: Probing the Oceans 1936 to 1976* (<https://web.archive.org/web/20190425203756/http://scilib.ucsd.edu/sio/hist/gc29s5.pdf>) (PDF). San Diego, California: Tofua Press. Archived from the original (<http://scilib.ucsd.edu/sio/hist/gc29s5.pdf>) (PDF) on 25 April 2019. Retrieved 19 April 2019.
22. Harmon Craig (26 May 1961). "Isotopic variations in meteoric waters". *Science*. **133** (3465): 1702–1703. Bibcode:1961Sci...133.1702C (<https://ui.adsabs.harvard.edu/abs/1961Sci...133.1702C>). doi:10.1126/science.133.3465.1702 (<https://doi.org/10.1126%2Fscience.133.3465.1702>). PMID 17814749 (<https://pubmed.ncbi.nlm.nih.gov/17814749>). S2CID 34373069 (<https://api.semanticscholar.org/CorpusID:34373069>).
23. Giggenbach, W.F. (November 1992). "Isotopic shifts in waters from geothermal and volcanic systems along convergent plate boundaries and their origin". *Earth and Planetary Science Letters*. **113** (4): 495–510. Bibcode:1992E&PSL.113..495G (<https://ui.adsabs.harvard.edu/abs/1992E&PSL.113..495G>). doi:10.1016/0012-821X(92)90127-H (<https://doi.org/10.1016%2F0012-821X%2892%2990127-H>).
24. Craig, H. (1963). "The isotopic geochemistry of water and carbon in geothermal areas". In Tongiorgi, E. (ed.). *Nuclear Geology on Geothermal Areas. Proceedings of the First Spoleto Conference, Spoleto, Italy*. Pisa: V. Lischi & Figli. pp. 17–53.
25. Craig, H.; Gordon, L.I. (1965). "Deuterium and oxygen 18 variations in the ocean and the marine atmosphere". In Tongiorgi, E. (ed.). *Stable Isotopes in Oceanographic Studies and Paleotemperatures, Proceedings of the Third Spoleto Conference, Spoleto, Italy*. Pisa: V. Lischi & Figli. pp. 9–130.
26. Craig, H. (1965). "The measurement of oxygen isotope paleotemperatures". In Tongiorgi, E. (ed.). *Stable Isotopes in Oceanographic Studies and Paleotemperatures, Proceedings of the Third Spoleto Conference, Spoleto, Italy*. Pisa: V. Lischi & Figli. pp. 161–182.
27. Strauch, Gerhard; Gonfiantini, Roberto (March 2008). "Another special issue: 40 years Craig–Gordon model of isotope fractionation of water". *Isotopes in Environmental and Health Studies*. **44** (1): 1. Bibcode:2008IEHS...44....1S (<https://ui.adsabs.harvard.edu/abs/2008IEHS...44....1S>). doi:10.1080/10256010801887000 (<https://doi.org/10.1080%2F10256010801887000>). PMID 18320422 (<https://pubmed.ncbi.nlm.nih.gov/18320422>). S2CID 34179724 (<https://api.semanticscholar.org/CorpusID:34179724>).
28. Craig, H.; Gordon, L. I.; Horibe, Y. (1 September 1963). "Isotopic exchange effects in the evaporation of water: 1. Low-temperature experimental results". *Journal of Geophysical Research*. **68** (17): 5079–5087. Bibcode:1963JGR....68.5079C (<https://ui.adsabs.harvard.edu/abs/1963JGR....68.5079C>). doi:10.1029/JZ068i017p05079 (<https://doi.org/10.1029%2FJZ068i017p05079>).
29. Jenkins, William J.; Doney, Scott C.; Fendrock, Michaela; Fine, Rana; Gamo, Toshitaka; Jean-Baptiste, Philippe; Key, Robert; Klein, Birgit; Lupton, John E.; Newton, Robert; Rhein, Monika; Roether, Wolfgang; Sano, Yuji; Schlitzer, Reiner; Schlosser, Peter; Swift, Jim (5 April 2019). "A comprehensive global oceanic dataset of helium isotope and tritium measurements" (<https://doi.org/10.5194%2Fessd-11-441-2019>). *Earth System Science Data*. **11** (2): 441–454. Bibcode:2019ESSD...11..441J (<https://ui.adsabs.harvard.edu/abs/2019ESSD...11..441J>). doi:10.5194/essd-11-441-2019 (<https://doi.org/10.5194%2Fessd-11-441-2019>).

30. Jenkins, W. J. (October 8, 2002). "W. Brian Clarke Professor Emeritus Physics & Astronomy 1937-2002" (<https://www.mcmaster.ca/mufa/newsOct02.html>). *McMaster University*. Retrieved 24 April 2019.
31. Emiliani, Cesare (June 28, 2005). *Oceanic lithosphere* (<https://books.google.com/books?id=gRZihrnakJoC&pg=PA394>). Harvard University Press. pp. 392–394. ISBN 9780674017368.
32. Clarke, W.B.; Beg, M.A.; Craig, Harmon (June 1969). "Excess ³He in the sea: Evidence for terrestrial primodal helium". *Earth and Planetary Science Letters*. **6** (3): 213–220. doi:10.1016/0012-821X(69)90093-4 (<https://doi.org/10.1016%2F0012-821X%2869%2990093-4>).
33. Nozaki, Yoshiyuki; Zhang, Jing; Takeda, Akihisa (January 1997). "²¹⁰Pb and ²¹⁰Po in the equatorial Pacific and the Bering Sea: the effects of biological productivity and boundary scavenging" (<https://www.researchgate.net/publication/222235022>). *Deep Sea Research Part II: Topical Studies in Oceanography*. **44** (9–10): 2203–2220. doi:10.1016/S0967-0645(97)00024-6 (<https://doi.org/10.1016%2FS0967-0645%2897%2900024-6>). Retrieved 24 April 2019.
34. Craig, H.; Krishnaswami, S.; Somayajulu, B.L.K. (January 1973). "²¹⁰Pb/²²⁶Ra: Radioactive disequilibrium in the deep sea". *Earth and Planetary Science Letters*. **17** (2): 295–305. doi:10.1016/0012-821X(73)90194-5 (<https://doi.org/10.1016%2F0012-821X%2873%2990194-5>).
35. Craig, H.; Chung, Y.; Fiadeiro, M. (September 1972). "A benthic front in the South Pacific". *Earth and Planetary Science Letters*. **16** (1): 50–65. Bibcode:1972E&PSL..16...50C (<https://ui.adsabs.harvard.edu/abs/1972E&PSL..16...50C>). doi:10.1016/0012-821X(72)90236-1 (<https://doi.org/10.1016%2F0012-821X%2872%2990236-1>).
36. King, Chi-Yu (10 November 1986). "Gas geochemistry applied to earthquake prediction: An overview". *Journal of Geophysical Research: Solid Earth*. **91** (B12): 12269–12281. Bibcode:1986JGR....9112269K (<https://ui.adsabs.harvard.edu/abs/1986JGR....9112269K>). doi:10.1029/JB091iB12p12269 (<https://doi.org/10.1029%2FJB091iB12p12269>).
37. Craig, H; Lupton, J E; Chung, Y; Horowitz, R M (1977). *Technical Report No.7, Investigation of radon and helium as possible fluid-phase precursors to earthquakes. Technical Report No. 2, Additional task: Radon, helium and geochemical monitoring on the Palmdale uplift* (<https://escholarship.org/uc/item/9q82h532>). La Jolla, California: Scripps Institution of Oceanography. Retrieved 19 April 2019.
38. Shapiro, M. H.; Melvin, J. D.; Tombrello, T. A.; Mendenhall, M. H.; Larson, P. B.; Whitcomb, J. H. (1981). "Relationship of the 1979 Southern California Radon Anomaly to a possible regional strain event" (<https://web.archive.org/web/20190425194532/https://pdfs.semanticscholar.org/ee58/595023dacad2b5df795d0f3deb3bb2e4d331.pdf>) (PDF). *Journal of Geophysical Research*. **86** (B3): 1725. Bibcode:1981JGR....86.1725S (<https://ui.adsabs.harvard.edu/abs/1981JGR....86.1725S>). doi:10.1029/JB086iB03p01725 (<https://doi.org/10.1029%2FJB086iB03p01725>). S2CID 40963868 (<https://api.semanticscholar.org/CorpusID:40963868>). Archived from the original (<https://pdfs.semanticscholar.org/ee58/595023dacad2b5df795d0f3deb3bb2e4d331.pdf>) (PDF) on 2019-04-25. Retrieved 25 April 2019.
39. Tambakopoulos, Dimitris; Stefanidou-Tiveriou, Theodosia; Papagianni, Eleni; Maniatis, Yannis (23 October 2017). "Provenance investigation of Roman marble sarcophagi from Nicopolis, Epirus, Greece: revealing a strong artistic and trade connection with Athens". *Archaeological and Anthropological Sciences*. **11** (2): 597–608. doi:10.1007/s12520-017-0556-8 (<https://doi.org/10.1007%2Fs12520-017-0556-8>). S2CID 134878848 (<https://api.semanticscholar.org/CorpusID:134878848>).
40. Herz, N. (April 17, 2013). "The oxygen and carbon isotopic data base for classical marble" (<https://books.google.com/books?id=Y3fyCAAQBAJ&pg=PA303>). In Herz, N.; Waelkens, Marc (eds.). *Classical Marble: Geochemistry, Technology, Trade*. Springer Science & Business Media. pp. 305–314. ISBN 9789401577953. Retrieved 25 April 2019.

41. Craig, H.; Craig, V. (28 April 1972). "Greek Marbles: Determination of Provenance by Isotopic Analysis". *Science*. **176** (4033): 401–403. Bibcode:1972Sci...176..401C (<https://ui.adsabs.harvard.edu/abs/1972Sci...176..401C>). doi:10.1126/science.176.4033.401 (<https://doi.org/10.1126%2Fscience.176.4033.401>). JSTOR 1734394 (<https://www.jstor.org/stable/1734394>). PMID 17777722 (<https://pubmed.ncbi.nlm.nih.gov/17777722>). S2CID 1833369 (<https://api.semanticscholar.org/CorpusID:1833369>).
42. Page, Douglas (1999). "Harmon Craig: The Gumshoe of Geochemistry" (<https://www.scribd.com/document/16298030/Harmon-Craig-Gumshoe-of-Geochemistry>). *Scribd*.
43. Lupton, John E.; Craig, Harmon (October 2, 1981). "A Major Helium-3 Source at 15° S on the East Pacific Rise". *Science, New Series*. **214** (4516): 13–18. doi:10.1126/science.214.4516.13 (<https://doi.org/10.1126%2Fscience.214.4516.13>). JSTOR 1687232 (<https://www.jstor.org/stable/1687232>). PMID 17802550 (<https://pubmed.ncbi.nlm.nih.gov/17802550>). S2CID 179027416 (<https://api.semanticscholar.org/CorpusID:179027416>).
44. Broecker, W.S. (September 29, 1980). "Chapter 15: Geochemical Tracers and Ocean Circulation" (https://ocw.mit.edu/resources/res-12-000-evolution-of-physical-oceanography-spring-2007/part-3/wunsch_chapter15.pdf) (PDF). In Warren, Bruce A.; Wunsch, Carl (eds.). *Evolution of Physical Oceanography*. The MIT Press. pp. 434–461. Retrieved 25 April 2019.
45. "Harmon Craig Wins Balzan Prize" (<https://www.geochemsoc.org/files/8313/4436/8118/gn098.pdf>) (PDF). *The Geochemical News*. No. January. 1999. p. 8. Retrieved 18 April 2019.
46. "Harmon Craig USA 1998 Balzan Prize for Geochemistry (Acceptance Speech – Rome)" (<https://www.balzan.org/en/prizewinners/harmon-craig/rome-23-11-1998-craig>). *International Balzan Prize Foundation*. November 23, 1998. Retrieved 18 April 2019.
47. Craig, H.; Chou, C. C. (November 1982). "Methane: The record in polar ice cores". *Geophysical Research Letters*. **9** (11): 1221–1224. Bibcode:1982GeoRL...9.1221C (<https://ui.adsabs.harvard.edu/abs/1982GeoRL...9.1221C>). doi:10.1029/GL009i011p01221 (<https://doi.org/10.1029%2FGL009i011p01221>).
48. Craig, H.; Horibe, Y.; Sowers, T. (23 December 1988). "Gravitational Separation of Gases and Isotopes in Polar Ice Caps". *Science*. **242** (4886): 1675–1678. Bibcode:1988Sci...242.1675C (<https://ui.adsabs.harvard.edu/abs/1988Sci...242.1675C>). doi:10.1126/science.242.4886.1675 (<https://doi.org/10.1126%2Fscience.242.4886.1675>). PMID 17730578 (<https://pubmed.ncbi.nlm.nih.gov/17730578>). S2CID 34912363 (<https://api.semanticscholar.org/CorpusID:34912363>).
49. "Two Geochemists Win Prizes in Earth Science" (<https://www.nytimes.com/1987/11/19/us/two-geochemists-win-prizes-in-earth-science.html>). *The New York Times*. November 19, 1987. Retrieved 25 April 2019.
50. Marti, Kurt; Weiss, Ray F.; Winterer, Edward L. "In Memoriam: Harmon Craig Professor of Oceanography" (https://senate.universityofcalifornia.edu/_files/inmemoriam/html/HarmonCraig.html). *UC San Diego*. UC San Diego Senate. Retrieved 25 April 2019.
51. "Harmon Craig" (<http://www.nasonline.org/member-directory/deceased-members/56655.html>). *National Academy of Sciences*. Retrieved 18 April 2019.
52. "Harmon Craig Biography" (<https://www.ldeo.columbia.edu/the-vetlesen-prize/past-recipients/harmon-craig>). *Lamont–Doherty Earth Observatory*.
53. "Harmon Craig wins Balzan Prize" (<https://web.archive.org/web/20190802123503/http://www.socarchsci.org/bulletin/9901/9901n.htm>). *The Society for Archaeological Sciences Bulletin*. Society for Archaeological Sciences. 1998. Archived from the original (<https://www.socarchsci.org/bulletin/9901/9901n.htm>) on 2 August 2019. Retrieved 18 April 2019.
54. Turekian, Karl K. (June 2003). "Harmon Craig (1926–2003)" (<https://doi.org/10.1038%2F423701a>). *Nature*. **423** (6941): 701. doi:10.1038/423701a (<https://doi.org/10.1038%2F423701a>). PMID 12802321 (<https://pubmed.ncbi.nlm.nih.gov/12802321>).

External links

- [Oral history interview transcript with Harmon Craig on 29 April 1996, American Institute of Physics, Niels Bohr Library & Archives \(https://www.aip.org/history-programs/niels-bohr-library/oral-histories/32508\)](https://www.aip.org/history-programs/niels-bohr-library/oral-histories/32508)
 - ["Accession No.: 2003-41 PROCESSING RECORD: Harmon Bushnell Craig Papers, 1948-2003" \(https://library.ucsd.edu/speccoll/findingaids/Craig2003-41.pdf\) \(PDF\). *SCRIPPS INSTITUTION OF OCEANOGRAPHY ARCHIVES*. Retrieved 19 April 2019.](https://library.ucsd.edu/speccoll/findingaids/Craig2003-41.pdf)
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